

Mathematical modeling is an integral component of the practice of synthetic biology: permitting the simulation of the behaviour of genetic circuits in biological systems. Mathematics could be harnessed to inform laboratory projects through predicting the characteristics and response of parts and systems *a priori* as well as to explain experimentally-derived data *post hoc*. Thus, our team developed a mathematical framework to predict the behaviour of elements of our design.

Models were developed for two main purposes:

1. Predicting the effects of changing extracellular conditions on promoter activity through a series of kinetic equations
2. Simulating the stability of residues within the acyl homo-L-serine (AHL) synthase protein structure in response to the binding of its substrate S-adenosyl-L-methionine (SAM) to elucidate the putative residues that participate in its active site

In the first model, we applied ordinary differential equations (ODEs) for transcription and translation with previously-reported parameters in the literature to predict the effect of extracellular pH on *gadA* promoter activity. In the second model, we utilized molecular dynamics simulations for SAM-binding determination of AHL synthase *in silico*.

#### Reaction Kinetics at the *gadA* pH-responsive Promoter

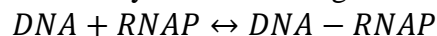
To predict the responsiveness of the pH promoter to changing extracellular pH within a heterogenous tumour microenvironment, we developed kinetic models to simulate the activity of the *gadA* promoter in the bacterial cell.

#### Method

In order to simulate the many mechanistic components of the pH-sensitive *gadA* promoter, we applied dissociation constants found in literature to a series of ODEs.

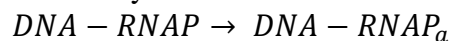
The behaviour of the *gadA* promoter was predicted using the framework of the basic model of bacterial transcription and translation at a promoter as suggested by the central dogma.

#### RNA Polymerase Binding:



$$\frac{d[DNA-RNAP]}{dt} = k_{aRNAP-DNA}[DNA][RNAP] - k_{dRNAP-DNA}[DNA - RNAP]$$

#### RNA Polymerase Activation:



$$\frac{d[DNA-RNAP_a]}{dt} = k_a[DNA - RNAP]$$

#### mRNA Transcription:



$$\frac{d[mRNA]}{dt} = k_{TX}[DNA - RNAP_a]$$

mRNA degradation:



$$\frac{d[mRNA]}{dt} = -k_{mRNAdeg}[mRNA]$$

Translation as modelled through Michaelis-Menten kinetics:



$$\frac{d[Protein]}{dt} = \frac{k_{TL}[Ribosome][mRNA]}{k_M + [mRNA]}$$

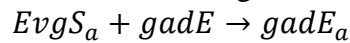
The *gadA* pH-responsive element can be understood most simply by the following transduction pathway as described by Foster (2004):

1. Detection from the membrane-bound sensor kinase EvgS
2. Activation of the *gadE* transcription factor
3. *gadE* binding to the *gadA* promoter element
4. *gadE*-induced transcription at the *gadA* locus

Detection of acidic protons with the membrane-bound sensor kinase EvgS

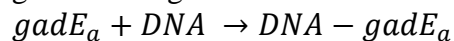


Activation of the *gadE* transcription factor

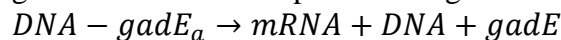


These aforementioned processes can be modelled through least-squares curve-fitting on a Lorentzian curve with previously reported data associating pH with transcriptional activity through EvgS Histidine Kinase activation (Eguchi, Utsumi, 2014). As Eguchi and Utsumi reported their data in Miller units, the curve was transformed as a measure of *gadE* activation through assuming *Miller units*  $\propto$  *gadE* activation, with a ceiling of  $gadE_{init} = 6.1 \times 10^{-10} M$ .

*gadE* binding



*gadE*-induced transcription at the *gadA* locus



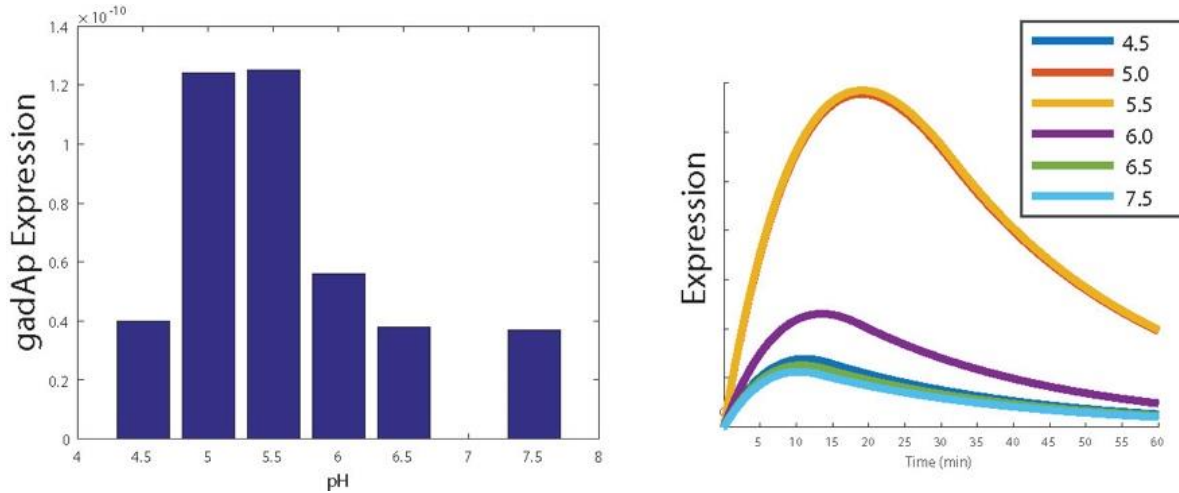
**Parameter values as reported by literature:**

Variable	Description	Estimated Value	Reference
$k_{aRNAP-DNA}$	RNA Polymerase Association constant with DNA	$5.7 \times 10^6 M^{-1} s^{-1}$	Bertrand-Burggraf et al., 1987

$k_{dRNAP-DNA}$	RNA Polymerase Dissociation constant with DNA	$10 s^{-1}$	Kierzek, Zaim & Zielenkiewicz, 2001
$k_a$	Closed complex isomerization	$10.5 \times 10^{-2} s^{-1}$	Bertrand-Burggraf et al., 1987
$k_{TX}$	Transcription kinetic constant	$1/300 s^{-1}$	Karzbrun et al., 2011
$k_{mRNAdeg}$	mRNA degradation kinetic constant	$0.3 s^{-1}$	Kierzek, Zaim & Zielenkiewicz, 2001
$k_{TL}$	Translation kinetic constant	$4/65 s^{-1}$	Karzbrun et al., 2011
$k_{dgadE}$	gadE dissociation constant	$6 \mu M$	Krin, Danchin & Soutourina, 2010
$gadE_{init}$	Initial concentration of gadE transcription factor	$6.1 \times 10^{-10} M =$ $\frac{370 \frac{gadE}{E\text{-coli Cell}}}{\frac{6.02 \times 10^{23} gadE/mol}{\frac{1 L}{1 dm^3} \frac{1 \mu m^3}{E\text{-coli Cell}} \frac{1 dm^3}{10^{12} \mu m^3}}}$	Ishihama et al., 2014 – estimation used Dan TF as a surrogate measure (part of the LysR TF family)
$DNA_{init}$	Initial concentration of DNA	$26 \times 10^{-6} M =$ $\frac{0.017 \frac{pg DNA}{E\text{-coli Cell}}}{\frac{650 g/mol}{\frac{1 L}{1 dm^3} \frac{1 \mu m^3}{E\text{-coli Cell}} \frac{1 dm^3}{10^{12} \mu m^3}}}$	ThermoFisher Scientific, nd
$RNAP_{init}$	Initial concentration of RNA polymerase	$2.5 \times 10^{-6} M$	Shepherd, Dennis, & Bremer, 2001
$Ribosome_{init}$	Initial concentration of cytoplasmic ribosomes	$50 \times 10^{-9} M =$ $\frac{3 \times 10^4 \frac{Ribosomes}{E\text{-coli Cell}}}{\frac{6.02 \times 10^{23} ribosomes/mol}{\frac{1 L}{1 dm^3} \frac{1 \mu m^3}{E\text{-coli Cell}} \frac{1 dm^3}{10^{12} \mu m^3}}}$	ThermoFisher Scientific, nd

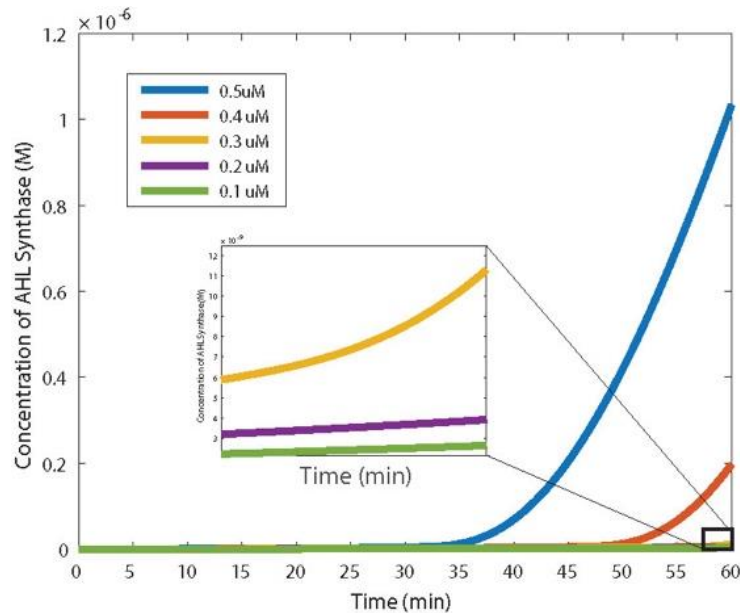
## Modelling *gadA* Activity in Response to a pH Series

Utilizing our previously described kinetic ODEs, we modelled the differential gene expression at the *gadA* promoter changes in response to changing extracellular pH. The following are the transformed experimental data from Eguchi & Utsumi (2014; left) utilized for curve-fitting and the simulated results (right):



## Modelling AHL Synthase Concentration in Response to Changing extracellular [H<sup>+</sup>]

As our genetic circuit placed AHL synthase downstream of the *gadA* promoter to initiate quorum sensing, we modelled how the intracellular concentration of AHL synthase would be altered due to altered activity of the *gadA* promoter to a series of changing extracellular pH (as indicated by [H<sup>+</sup>]). We thus decided to parse the behaviour of AHL synthase expression in a [H<sup>+</sup>] series. following curves are the simulated results:



Our model predicts that 0.5 uM of [H+] elicited the strongest induction of AHL synthase expression. The results of our model are in accordance with previously-determined experimental data characterizing the gadA promoter (link Dundee 2016 wiki page: <http://2016.igem.org/Team:Dundee/Result>). The results of this kinetic model affirm that the optimal activity of the gadA promoter lies between pH 5-5.5. This is concordant with the pH range of the tumour microenvironment as characterized by imaging studies (Chen & Pagel, 2015). For example, Castelli and colleagues (2014) determined that the *in vivo* extracellular pH of murine melanoma ranged from 5.2–6.4. This suggests that the gadA promoter can be applied to our project, which aims to develop a self-limiting tumour-killing genetic circuit responsive to an acidic tumour microenvironment.

## References:

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