

Conjugation of antigen to gold nanoparticles

A. Preparation :

- Measure the concentration of antigen at O.D. 280 nm
- Know the quantity of the final product
- Know the pH of the final product (adjusted by K_2CO_3)

B. Theory :

Gold nanoparticles tend to conjugate on antigen due to the van der Waals force

C. Material & Method :

- Prepare 400 μ l 5X gold nanoparticle buffer, diluted by 2mM borax buffer and filtered by 0.22 μ m filter
- add ___ μ g antibody (diluted 10X in 0.01M PBS), add into the solution in step1
- RT stir 5~6 rpm , 60 min (particles are shown in bright red)
- Add 10%BSA to 0.01M PBS , filter through 0.45 μ m filter , add solution 220 μ L to the gold solution , the final concentration will be 1% BSA. RT stir for 30 minutes
- Distribute the solution into 2 tubes , centrifuge at 4 $^{\circ}$ C , 13000rpm for 30 minutes
- Remove the supernatant , restore the pellet by buffer A, 90 μ L each
- Store the conjugated antigen in a 4ml vial, 4 $^{\circ}$ C freezer
- Final solution are shown in burgundy red, results : I is larger than 20ppb~50ppb